# The Impact of Silver Nanoparticles on Plant Biomass and Chlorophyll Content

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**ABSTRACT:** Applications of nanoparticles by industry and its disposal is a new concerned for environment. Depending upon the concentration of particles and its exposure time causes negative impact on V.radiata and B.campestris seedlings. V.radiata was the only species among two test plants which was found to be resistance to Ag nanoparticle and ion solution. Significant inhibition on shoot fresh weight of V.radiata (p=0.008) and B.campestris (p=0.002) was observed at 1000 µg/mL silver nanoparticle solution after treatment period. V.radiata showed significant retardation on dry weight of root at 1000 µg/mL of Ag<sup>+</sup> ions solution after 12<sup>th</sup> day. The decrease on shoot dry weight with increase in nanoparticle and ion concentration was also observed after 12<sup>th</sup> day. Exposure to 1000 µg/mL of Ag nanoparticles reported significant retardation on total chlorophyll content V.radiata (p=0.001) and B.campestris (p=0.001) when compare to control after 12<sup>th</sup> day of treatment. After the treatment period no significant inhibition on chlorophyll ratio was observed when exposed to both Ag nanoparticle and ion solutions. Transmission Electron Microscope reveals breakage of cell wall and vacuoles of test plants which shows the toxic nature of Ag nanoparticles inside treated root cells.

Key words: Chlorophyll, Concentration, Nanoparticles, Toxic and Transmission Electron Microscope (TEM).

# I. INTRODUCTION:

Nanotechnology requires control and construction of improved new material at nanoscale level in which integration of nanoscale structures into larger material components and systems take place [1]. Both positive and negative impacts of nanoparticles on higher plants were reported. Nano-SiO<sub>2</sub> and nano-TiO<sub>2</sub> caused an increase in nitrate reductase in *Glycine max* which enhances the ability to absorb and utilize water. However, these two nanoparticles stimulated antioxidant systems and hastened the germination and growth of plants [2]. Effect of silver nanoparticles on reduction of biomass and transpiration rate was also reported in *Cucurbita* pepo. The adverse effect on C. pepo was more prevalent in nanoparticles than bulk silver solutions (4.4 to 10 times more) [3]. Magnetic nanoparticles coated with stabilizers such as Tetramethylammonium hydroxide (TMA-OH) was studied on early growth stages of maize plants. Small concentration of aqueous ferrofluid solution added in a culture medium had a stimulating effect on the growth of plants while the enhanced concentration of aqueous ferrofluid solution induced an inhibitory effect. It was found that at low concentration of ferrofluid, there was an increase in chlorophyll "a" level while at higher concentration it was inhibited [4]. Copper nanoparticles of higher concentration (1000 mg/L) caused adverse effect on seedling growth of mung bean. TEM images showed that particles were mostly deposited at 1000 mg/L than 200 mg/L. Copper nanoparticles crossed the cell membrane of P. radiatus and T. aestivum and aggregated along with the other cellular materials within the cells. The dispersions of nanoparticles resulted in no precipitations in culture plate in this new technique [5]. Bioaccumulation of nanoparticles increased with increase in concentrations of growth media and their bioavailability to test plant was calculated by the bioaccumulation factor. Effect of copper nanoparticles on zucchini plants showed inhibition of root length in seedling compare to control [6].

## II. MATERIALS AND METHOD:

#### 2.1 Synthesis of Ag nanoparticles solution:

Silver nanoparticles were synthesized in the aqueous phase, using double distilled water. All reagents were purchased from Merck chemicals and used as received. For Ag nanoparticles preparations,  $10^{-3}$  M AgNO<sub>3</sub> solutions were reduced with  $10^{-3}$  M NaBH<sub>4</sub> in double distilled water. Tween-20 was added as a surfactant to prevent aggregation of particles. Silver ion solutions were prepared in double distilled water in absence of NaBH<sub>4</sub> and Tween-20 [7].

#### 2.2 Seedling growth:

*V.radiata* and *B.campestris* seeds were selected for the study. The seeds were germinated and uniform seedlings were selected for experiments. The seedlings were grow in Hoagland nutrient solution and transferred in different concentration of nanoparticles and ion solutions at growth chamber. The Phytotoxicity periods continue for 12 days [7].

#### 2.3 Fresh and dry matter estimation:

Fresh weight was measured at different intervals. The treated and untreated seedlings were washed under tap water and then rinsed in distilled water. Roots and shoots were separated and blotted dry. Dry weight was measured by drying root and shoot at 70  $^{\circ}$ C for 24 hours in an oven [8].

## 2.4 Quantifications of Chlorophyll.

Chlorophyll a, b and total chlorophyll was measured by extracting 0.5g of fresh leaf in 3 mL of 80% acetone with a small amount of quartz sand. The homogenate was filtered through Whatman No.1 filter paper. The color intensity was measured at 645 nm and 663 nm using UV-Vis spectrophotometer (Hitachi, Model no.3210).

# 2.5 Transmission Electron Microscope observations:

Localization of nanoparticles was studied using TEM (Model - JEOL JSM 100 CX). At first segments were taken from the treated seedling roots above the apical part of the root tip. The sample was prepared by standard procedure followed at Sophisticated Analytical Instrument Facility, NEHU, Shillong.

#### 1.6 Statistical analysis:

In every experiment, each treatment was conducted with three replicates. The statistical analysis of experimental values was compared with the control. Statistical significance was done by student–t test analysis. It was accepted when the probability of the result by assuming null hypothesis (p) is less than 0.05.

# III. EXPERIMENTAL FINDING

# 3.1 Changes in fresh weight of treated and untreated test plants:

The test plants exhibited increase in biomass but at variable rate which depends on plant species, concentration of Ag nanoparticle and ions and its exposure time. No significant reduction on root fresh weight was reported after 1<sup>st</sup> day of treatment on *V.radiata* and *B.campestris* by Ag nanoparticle and ion solution. No significant inhibition on root fresh weight was observed at 50µg/mL and 500 µg/mL of Ag nanoparticle solution in test plants. Concentration of Ag ions (50 µg/mL, 500 µg/mL and 1000 µg/mL) exposed to test plants did not show significant inhibition on root fresh weight till 3<sup>rd</sup> day. Similar result was obtained when ZnO nanoparticle (1 ppm and 20 ppm) resulted in an increase in root and shoot biomass of mung and gram seedling. The increase in biomass at 1 ppm and 20 ppm concentration suggests the optimum dose limit for the growth of mung and gram seedlings [9].

Adverse effects on root fresh weight was observed from 6<sup>th</sup> day onwards by both nanoparticle and ion treatment. Effect on root fresh weight was observed beyond 50 µg/mL of Ag nanoparticle concentration in test plants. Significant reduction on root fresh weight was observed at 500 µg/mL in *V.radiata* (p=0.027) and *B.campestris* (p=0.024) compared to control after 6<sup>th</sup> day. However 1000 µg/mL of Ag nanoparticle solution reported adverse effect on root fresh weight of, *V.radiata* (p=0.005) and *B.campestris* (p=0.013) compared to control. Ag<sup>+</sup> ion solution showed significant inhibition at 500 µg/mL on fresh weight of root in *B.campestris* (p=0.027). 1000 µg/mL of Ag<sup>+</sup> ion showed significant retardation on fresh weight of root in *V.radiata* (p=0.012) and *B.campestris* (p=0.006) compared to control. There was a significant reduction on root fresh weight at 500 µg/mL and 1000 µg/mL of Ag nanoparticle solution in *B.campestris* plants. At 1000 µg/mL Ag<sup>+</sup> ion solution, significant effect was observed in *B.campestris* at 50 µg/mL, 500 µg/mL and 1000 µg/mL concentration of silver ions after 9<sup>th</sup> day. *V.radiata* was the only species among two test plants which was found to be resistance to Ag nanoparticle and ion solution after 12<sup>th</sup> day (Fig.1). However, *B.campestris* reported adverse effect on root fresh weight at 500 µg/mL and 1000 µg/mL for the formation of the format

The Fresh weight of shoot in test plants remained unaffected by the experimental values. No significant effect was observed on shoot fresh weight of test plants by both Ag nanoparticle and ion solution. However, the influence on shoot fresh weight was clearly observed from  $3^{rd}$  day by both Ag nanoparticle and ion solutions. Similar to the patterns of root fresh weight, the shoot fresh weight was also not affected at low concentration i.e. 50 µg/mL of Ag nanoparticle and ion solution after  $3^{rd}$  day. *B.campestris* showed significant retardation on shoot fresh weight at both 500 µg/mL (p=0.048) and 1000 µg/mL (p=0.017) of Ag nanoparticle solution after  $3^{rd}$  day. *B.campestris* showed significant inhibition on shoot fresh weight at 1000 µg/mL (p=0.015) of Ag nanoparticle solution after  $3^{rd}$  day. Among Ag<sup>+</sup> ion solution, 1000 µg/mL showed significant inhibition on shoot fresh weight in *V.radiata* (p=0.043) when compared with control.



Fig 1: Effect of Ag nanoparticles and *V.radiata* and *B.campestris* biomass (fresh weight) after 12 days of treatment.

Adverse effect on shoot fresh weight was observed from 6<sup>th</sup> day onwards. It was found that shoot fresh weight of *V.radiata* shows resistance to both Ag nanoparticle and ion treatments. *B.campestris* showed significant inhibition on shoot fresh weight at 500 µg/mL (p=0.001) and 1000 µg/mL (p=0.000) of Ag nanoparticle solutions when compared with control. Significant retardation on shoot fresh weight was also observed in *B.campestris* (p=0.012) by 1000 µg/mL of Ag<sup>+</sup> ion solution compare to control after 6<sup>th</sup> day. *B.campestris* shoot fresh weight was significantly inhibited after 9<sup>th</sup> day at 500 µg/mL and 1000 µg/mL of Ag nanoparticle solution. Significant retardation was observed in *V.radiata* (p=0.026) and *B.campestris* (p=0.030) at 1000 µg/mL of Ag<sup>+</sup> ion solution after 9<sup>th</sup> day. Significant inhibition on shoot fresh weight of *V.radiata* (p=0.008) and *B.campestris* (p=0.002) was observed at 1000 µg/mL Ag nanoparticle solution compared to control after 12<sup>th</sup> day of treatment (Fig.1). Similar result was obtained when ZnO nanoparticle of 2000 ppm caused decrease in biomass of mung and Gram seedling. Decrease in biomass of root and shoot shows the toxic nature of ZnO nanoparticle beyond 20 ppm concentration [9]. It was reported that decreased in fresh weight by silver ion was probably due to increase in metabolic activities in sunflower seedling [10]. Adverse effect of silver ions on fresh weight of sunflower plant supports our results since there was a decline in fresh weight of root and shoot by both Ag nanoparticle and ion solution during treatment period.

3.2 Changes in dry weight of treated and untreated test plants:

The influence of Ag nanoparticle and ion on dry weight of root of test plants after treatment period was shown in Fig.2. Dry weight production of root of test plants was not inhibited by any concentration of the Ag nanoparticle and ion treatment on 1<sup>st</sup> day. On the contrary, the test plants exhibited increase in dry weight with time, but at different time rate. V. radiata showed significant retardation on dry weight of root at 1000 µg/mL of Ag nanoparticle solution after 3<sup>rd</sup> day of treatment. No significant inhibition on root dry weight was observed in B.campestris at 500 µg/mL (p=0.110) and 1000 µg/mL (p=0.102) of Ag nanoparticle solutions when compared to control. There was no adverse effect on root dry weight in V.radiata and B.campestris seedling by Ag<sup>+</sup> ion solution after 3<sup>rd</sup> day. Significant inhibition on dry weight of root was reported in *B.campestris* at 500 µg/mL and 1000  $\mu$ g/mL of Ag nanoparticle solution. The dry weights of roots of test plants were adversely affected by all concentration (50  $\mu$ g/mL, 500  $\mu$ g/mL and 1000  $\mu$ g/mL) of Ag nanoparticle solution after 9<sup>th</sup> day. Dry weight of *B.campestris* root showed significant inhibition at 50  $\mu$ g/mL (*p*=0.029), 500  $\mu$ g/mL (*p*=0.004) and 1000 µg/mL (p=0.000) of nanoparticle concentrations compare to control. Similarly V.radiata resulted in significant retardation of root dry weight at 50 µg/mL (p=0.002), 500 µg/mL (p=0.001) and 1000 µg/mL (p=0.000) of nanoparticle solutions when compare with control after 9<sup>th</sup> day of treatment. V.radiata showed significant retardation on dry weight of root at 50  $\mu$ g/mL (p=0.050), 500  $\mu$ g/mL (p=0.008) and 1000  $\mu$ g/mL (p=0.003) of Ag nanoparticles solution when compared to control after  $12^{\text{th}}$  day. *B.campestris* also reported significant inhibition at 50 µg/mL (*p*=0.037), 500 µg/mL (*p*=0.002) and 1000 µg/mL (*p*=0.000) on dry weight of root compared to control. V. radiata showed significant retardation on dry weight of root at 1000 µg/mL of Ag<sup>+</sup> ions solution after 12<sup>th</sup> day.

Fig.2 shows the dry weight of shoot of test plants as affected by the application of various concentration of Ag nanoparticle and ion solution after the treatment period. No significant effect on dry weight

of shoot was observed in all test plants after 1<sup>st</sup> day of treatment. Ag nanoparticle concentration beyond 50  $\mu$ g/mL resulted in significant inhibition on shoot dry weight in *V.radiata* and *B.campestris* after 3<sup>rd</sup> day. 500  $\mu$ g/mL of Ag nanoparticle solution showed significant retardation in *V.radiata* (p=0.026) and *B.campestris* (p=0.007) shoot dry weight compare to control. 1000  $\mu$ g/mL of Ag nanoparticle solution resulted in significant retardation on shoots dry weight in *V.radiata* (p=0.005) and *B.campestris* (p=0.002) when compare with control after 3<sup>rd</sup> day. Biomass reduction was more observed after 6<sup>th</sup> day in *B.campestris* compared to *V.radiata* when exposed to Ag nanoparticle solution. 50  $\mu$ g/mL (p=0.004), 500  $\mu$ g/mL (p=0.001) and 1000  $\mu$ g/mL (p=0.000) of Ag nanoparticle solution showed significant inhibition on dry weight of *B.campestris* shoot after 6<sup>th</sup> day of treatment. The test plants exhibited increase or decrease in shoot dry weight with exposure time and concentration. Significant retardation on shoot dry weight was observed at 1000  $\mu$ g/mL of Ag nanoparticle in *B.campestris* seedling after 9<sup>th</sup> day.



Fig 2: Effect of Ag nanoparticles and ions on *V.radiata* and *B.campestris* biomass (dry weight) after 12 days of treatment.

The decrease on shoot dry weight with increase in Ag nanoparticle and ion concentration was also observed after  $12^{\text{th}}$  day.  $500\mu\text{g/mL}$  (p=0.012) and  $1000 \ \mu\text{g/mL}$  (p=0.004) of nanoparticle solution showed significant retardation on dry weight of *V.radiata* shoot compared to control after  $12^{\text{th}}$  day. Significant changes on dry weight of *Lemna minor* L at different Ag nanoparticle concentrations are due to different nanoparticle size [11]. Similar result was also observed when 1000  $\mu\text{g/mL}$  nanoparticle concentration showed more adverse effect on dry weight of test plants than 50  $\mu\text{g/mL}$  and 500  $\mu\text{g/mL}$  Ag nanoparticle solution.

### 3.3 Estimation on chlorophyll content of seedlings:

Effect on total chlorophyll content exposed to different concentration of Ag nanoparticle and ion solutions was observed in V.radiata and B.campestris plants (Fig. 3A and Fig. 3B). The chlorophyll content of different crop plants was tolerant to both Ag nanoparticle and ion concentration used and, therefore, chlorophyll production was not affected till 3<sup>rd</sup> day. It was reported that Ag nanoparticles of 20 nm taken up by plants which were mostly in intracellular spaces could be transported inside plant cells through plasmadesmata of root cells [12]. These nanoparticles were then pass through shoots and then accumulated on leaves which caused adverse effect on total chlorophyll content of test plants. The total chlorophyll contents at 50 µg/mL of both Ag nanoparticle and ion solution did not showed any significant inhibition in test plants. A study reported that chlorophyll content of maize plants was found to be increased by low concentration (10-50 µl/L) while it was found to be inhibited by higher concentrations of magnetic nanoparticle [13]. B. campestris showed significant inhibition on total chlorophyll content at 500 µg/mL (p=0.044) and 1000 µg/mL (p=0.018) of Ag nanoparticle solutions compared to control. However V. radiata showed significant inhibition on total chlorophyll content at 1000  $\mu$ g/mL (p=0.017) of Ag nanoparticle solutions compared to control. 1000  $\mu$ g/mL (p=0.024) of Ag<sup>+</sup> ion solutions' showed significant retardation on total chlorophyll content of B.campestris. Adverse effect was observed at 1000  $\mu$ g/mL of Ag nanoparticles in V.radiata (p=0.033) and B.campestris (p=0.010) on total chlorophyll content compare to control. Higher concentration of Ag<sup>+</sup> ion i.e. 1000  $\mu$ g/mL resulted in significant inhibition in total chlorophyll content of V. radiata (p=0.002) and B. campestris (p=0.001) when compared to

control after 9<sup>th</sup> day. Moreover after 9<sup>th</sup> day of treatment Ag<sup>+</sup> ion solution showed significant inhibition in total chlorophyll content of *V.radiata* (p=0.038) compare to control. Increase in concentration of Ag nanoparticle showed significant effect on *V.radiata* and *B. campestris* after 12<sup>th</sup> day. It was observed that 500 µg/mL of nanoparticle solution shows significant retardation on total chlorophyll content in *V.radiata* (p=0.050) and *B.campestris* (p=0.010) compare to control. Exposure to 1000 µg/mL of Ag nanoparticles reported significant retardation on total chlorophyll content *V.radiata* (p=0.001) when compare to control after 12<sup>th</sup> day of treatment. *V.radiata* (p=0.001) and *B.campestris* (p=0.001) when compare to control after 12<sup>th</sup> day of treatment. *V.radiata* showed significant inhibition on total chlorophyll content at 500 µg/mL (p=0.012) and 1000 µg/mL (p=0.002) Ag<sup>+</sup> ion solution when compared to control

Fig. 4 shows the effect on chlorophyll ratio of test plants by both Ag nanoparticle and ion solutions. Decrease in chlorophyll ratio was observed in *B.campestris* with increase in concentration of nanoparticle solution. 500  $\mu$ g/mL of Ag<sup>+</sup> ion solution resulted in high chlorophyll ratio in *V.radiata* plants compared to 50  $\mu$ g/mL and 1000  $\mu$ g/mL Ag<sup>+</sup> ion concentrations. Increased in chlorophyll ratio was observed with increase in nanoparticle concentration in *V.radiata* after 3<sup>rd</sup> day. However *B.campestris* showed a decrease in chlorophyll ratio after 3<sup>rd</sup> day at 1000  $\mu$ g/mL (1.859±0.078) compared to 500  $\mu$ g/mL (1.918±0.102) Ag nanoparticle solutions. In *V.radiata*, 500  $\mu$ g/mL of Ag<sup>+</sup> ion solution showed a decrease in chlorophyll ratio compared to 50  $\mu$ g/mL and 1000  $\mu$ g/mL of Ag<sup>+</sup> ion solutions. It was observed that chlorophyll ratio decrease with increase in concentration of Ag<sup>+</sup> ion solutions. However *B.campestris* showed an increase in chlorophyll ratio with increase in concentration of Ag<sup>+</sup> ion solutions. However *B.campestris* showed an increase in chlorophyll ratio with increase in concentration of Ag<sup>+</sup> ion solutions. However *B.campestris* showed an increase in chlorophyll ratio with increase in concentration of Ag<sup>+</sup> ion solutions. However *B.campestris* showed an increase in chlorophyll ratio with increase in concentration of Ag<sup>+</sup> ion solutions. However *B.campestris* showed an increase in chlorophyll ratio with increase in Ag<sup>+</sup> ion concentrations after 3rd day. Chlorophyll ratio was found to be decreased with increase in Ag<sup>+</sup> ion concentration in *V.radiata* seedlings after 6<sup>th</sup> day.



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Fig 3: Effect of Ag nanoparticles and ions on total chlorophyll content of (A) *V.radiata* and (B) *B.campestris* in Hoagland nutrient solution during 12 days of treatment.



Fig 4: Effect of Ag nanoparticles and ions on chlorophyll ratio of *V.radiata* and *B.campestris* seedlings after 12 days of treatment.

*B.campestris* showed a decrease in chlorophyll ratio after 9<sup>th</sup> day at 1000 µg/mL compared to 50 µg/mL and 500 µg/mL of  $Ag^+$  ion solutions. However after  $12^{th}$  day of the treatment period no significant inhibition on chlorophyll ratio was observed on exposure to both Ag nanoparticle and ion solutions. Increase in LHC II content help to promotes energy transfer and oxygen evolutions in photosystem II in spinach [14]. It was also reported that increase in Hill reactions and activity on chloroplasts by nano-TiO<sub>2</sub> resulted in an acceleration of FeCy reduction and oxygen evolution in *Spinacia oleracea*. [15, 16]. Thus we can assume that Ag nanoparticle and ion solution at higher concentration (500 µg/mL and 1000 µg/mL) may directly affect the LHC II content on thylakoid membrane of selected test plants. This will increase the Hill reactions and activity on chloroplasts of *V.radiata* and *B.campestris* leaves.

# **3.4 TEM observations:**

Fig. 5 and Fig. 6 show detection of silver nanoparticles inside the root tissue of both *V.radiata* and *B.campestris*. The observation from the micrographic image (Fig.6) indicated that the whole cell and its intracellular portion i.e. plasmadesmata have silver nanoparticle particle. Magnified image of whole cell showed presence of individual and aggregated Ag particles which were clearly visible inside the cytoplasm of cell.





**Fig 5:** TEM images of the roots of *V*.*radiata* exposed to Ag nanoparticle of 1000  $\mu$ g/mL showing (A) Depositions of nanoparticles inside the cell, (B) Ag nanoparticles inside vacuoles.





B

**Fig 6:** TEM images of the roots of *B.campestris* exposed to Ag nanoparticle of 1000  $\mu$ g/mL showing (A)Deposition of nanoparticle inside whole cell, (B) Magnified portion of image A showing accumulation of particles in plasmadesmata and cell wall.

Accumulation of Ag nanoparticle was clearly observed inside vacuoles of root cell (Fig.5). Deposition of both individual and aggregate particle was found inside the cell wall which indicates the penetration of Ag particle inside the cells. The diameter of Ag nanoparticles was measured inside the plant cell and was found to be around 20 nm in size. The nanomaterials were found to be spherical in shape. One important hypothesis was established regarding transportation of smaller particles inside the cells. Cell walls thickness of about 5 to 20 nm functions act as natural sieves which transports small nanoparticles passes through large pores to enter in the protoplasm. New and large pores were created for passaging of larger nanoparticles at the cell wall [17, 18].

## **IV. CONCLUSION:**

Research on nanoparticles has received a great deal of interest in every discipline. Its applications can be found in many areas due to its high demand. But its adverse effect is always a concern for our environment. It penetrates easily inside the plant cells and causes effects on biomass and chlorophyll content. Deposition of small size nanoparticles inside the cell wall and vacuoles causes disturbance in metabolic activity of plants. The effect can be minimizing by limiting the concentration of nanoparticles solution used in different activity. *V.radiata* and *B.campestris* were both economically important plants, nanoparticles can easily find their way in human body through food chain. More investigations are needed to determine the negative impact of nanoparticles on crop plants and its consequences in other living organisms.

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#### **References:**

- [1] Y Ju-Nam, and J.R. Lead, Manufactured nanoparticles: an overview of their chemistry, interactions and potential environmental implications, *Sci Total Environ.*, 400 (1-3),2008, 396-414.
- [2] C.M. Lu, C.Y. Zhang, J.Q.Wen, G.R. Wu, and M.X. Tao, Research of the effect of nanometer materials on germinations and growth enhancement of Glycine max and its mechanism. *Soybean Sci.*, 21, 2002, 168-172.
- [3] C.Musante, and J. C. White, Toxicity of Silver and Copper to *Cucurbita pepo*: Differential Effects of Nano and Bulk-Size Particles, *Environ Toxicol.*, 2011, doi: 10.1002/tox.20667.

- M. Racuciu, and D.E. Creanga, TMA-OH Coated magnetic nanoparticles internalized in vegetal tissue. Rom. Journ. Phys., 52 (3-4), 2007, 395–402.
- [5] W.M. Lee, Y.J. An, H. Yoon, and H.S. Kweon, Toxicity and bioavailability of copper nanoparticles to the terrestrial plants Mung Bean (*Phaseolus radiatus*) and Wheat (*Triticum aestivum*): Plant agar test for water-insoluble nanoparticles. *Environ. Toxicol. Chem.*, 27 (9), 2008, 1915–1921.
- [6] D. Stampoulis, S.K. Sinha, and J.C. White, Assay-Dependent Phytotoxicity of Nanoparticles to Plants, *Environ. Sci. Technol.*, 43, 2009, 9473–9479.
- H. Mazumdar, and G.U.Ahmed, Phytotoxicity effect of silver nanoparticles on Oryza sativa, International Journal of Chemtech Research, 3 (3), 2011, 1494-1500.
- [8] D. Lin, and B. Xing, Root uptake and phytotoxicity of ZnO nanoparticles, *Environmental Science & Technology*, 42 (15), 2008, 5580-5585.
- [9] P. Mahajan, S.K. Dhoke, and A.S. Khanna, Effect of Nano-ZnO Particle Suspension on Growth of Mung (Vigna radiata) and Gram (Cicer arietinum) Seedlings Using Plant Agar Method, Journal of Nanotechnology, 2011, 7. doi:10.1155/2011/696535.
- [10] S. Krizkova, P. Ryant, O. Krystofova, V. Adam, M. Galiova, M. Beklova, P. Babula, J. Kaiser, K. Novotny, J. Novotny, M. Liska, R. Malina, J. Zehnalek, J. Hubalek, L. Havel, and R. Kizek, Multi-instrumental Analysis of Tissues of Sunflower Plants Treated with Silver(I) Ions Plants as Bioindicators of Environmental Pollution, *Sensors*, 8, 2008, 445-463.
- [11] E.J. Gubbins, L.C. Batty, and J.R. Lead, Phytotoxicity of silver nanoparticles to *Lemna minor* L. *Environmental Pollution*, *159*, 2011, 1551-1559.
- [12] X. Ma, J.G. Lee, Y. Deng, and A. Kolmakov, Interactions between engineered nanoparticles (ENPs) and plants: Phytotoxicity, uptake and accumulation, *Science of the Total Environment*, 408, 2010, 3053-3061.
- [13] M. Racuciu, and D.E. Creanga, TMA-OH Coated magnetic nanoparticles internalized in vegetal tissue. Rom. Journ. Phys., 52 (3-4), 2007, 395-402.
- [14] M. Su, F. Hong, C. Liu, X. Wu, X. Liu,L. Chen, F. Gao, F. Yang, and Z. Li, Effects of nano-anatase TiO<sub>2</sub> on absorption, distribution of light and photoreduction activities of chloroplast membrane of spinach, *Biol. Trace Elem. Res.*, 131 (1), 2009, 101.
- [15] F. Hong, J. Zhou, C. Liu, F. Yang, C. Wu, L. Zheng, and P. Yang, Effect of nano-TiO<sub>2</sub> on photochemical reaction of chloroplasts of spinach. *Biol. Trace Elem. Res.*, 105 (1-3), 2005a, 269–279.
- [16] F. Hong, F. Yang, C. Liu, Q. Gao, Z. Wan, F. Gu, C. Wu, Z. Ma, J. Zhou, and P. Yang, Influence of nano-TiO2 on the chloroplast aging of spinach under light. *Biological Trace Element Research*, 104, 2005b, 249-260.
- [17] E. Navarro, A. Baun, R. Behra, N.B. Hartmann, J. Filser, A.J. Miao, A. Quigg, P.H. Santschi, and L. Sigg, Environmental behaviour and ecotoxicology of engineered nanoparticles to algae, plant and fungi. *Ecotoxicology*, *17*, 2008a, 372-386
- [18] E. Navarro, F. Piccipetra, B. Wagner, F. Marconi, R. Kaegi, N. Odzak, L. Sigg, and R. Behra, Toxicity of silver nanoparticles to Chlamydomonas reinhardtii. Environmental Science and Technology, 42 (23), 2008b, 8959-8964.